

# Markscheme

November 2017

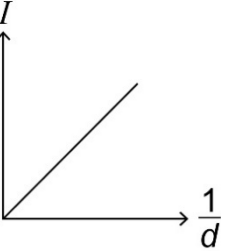
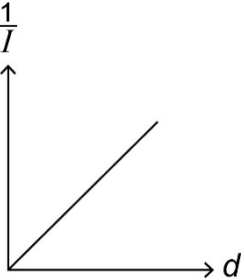
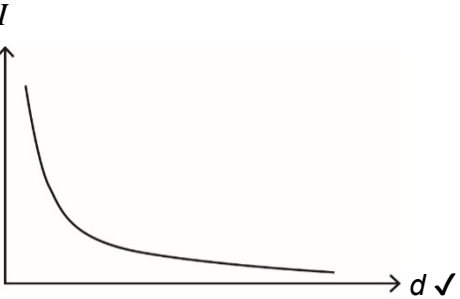
Chemistry

Standard level

Paper 3

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Section A

Question		Answers	Notes	Total
1.	a	<p><math>I</math></p>  <p>OR</p>  <p>OR</p> 	<p>Correct labels of axes required for mark.</p> <p>Accept <math>d^{-1}</math> instead of <math>\frac{1}{d}</math>.</p> <p>Accept <math>I^{-1}</math> instead of <math>\frac{1}{I}</math>.</p> <p>Plot of <math>I</math> vs <math>d</math> should not be linear.</p>	<p>1</p>

Question			Answers	Notes	Total
1.	b	i	negative correlation <b>OR</b> model/prediction matches results <b>OR</b> 99% of variance accounted for ✓		1
1.	b	ii	$I = -0.001631 d + 0.09939$ <b>OR</b> $y = -0.001631 x + 0.09939$ ✓	<i>Accept correctly rounded values for m and b in equation.</i> <i>Do not accept "y = mx + b".</i>	1
1.	b	iii	ions move «across electrolyte» ✓		1

Question		Answers	Notes	Total
2.	a	$\text{Mg(OH)}_2(\text{s}) + 2\text{HCl}(\text{aq}) \rightarrow \text{MgCl}_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l}) \checkmark$	Accept full or net ionic equation.	1
2.	b	<p>Any two from:</p> <p>volume «of HCl» <math>\checkmark</math></p> <p>concentration «of HCl»/[HCl] <math>\checkmark</math></p> <p>temperature «of HCl» <math>\checkmark</math></p> <p>mass of antacid/tablets <math>\checkmark</math></p> <p>size of antacid particles/tablets</p> <p><b>OR</b></p> <p>surface area of antacid «particles»/tablets <math>\checkmark</math></p>	<p>Accept “number of tablets/different doses”.</p> <p>Do <b>not</b> accept “same pH meter” <b>OR</b> “initial pH” <b>OR</b> “concentration of antacid/[antacid]”.</p> <p>A variable must be given so do <b>not</b> accept answers such as “stirring”, “whether tablets are whole or crushed” etc.</p>	2 max
2.	c	<p>(<math>\pm</math>) 0.04</p> <p><b>OR</b></p> <p>(<math>\pm</math>) 0.03 <math>\checkmark</math></p>		1

Question		Answers	Notes	Total
2.	d	<p>Any two of:</p> <p>uncertainty «(±)0.04/(±)0.03» means <b>A</b> and <b>C</b> cannot be distinguished ✓</p> <p>each measurement was conducted once ✓</p> <p>stomach pH should not be raised a lot «so antacid B is not necessarily effective» ✓</p> <p>mass/number of tablets/dose «of antacid» used was not controlled ✓</p> <p>actual environment in stomach is different ✓</p>	<p>Accept “amount of tablets” for “dose”.</p> <p>Do <b>not</b> accept “nature/composition of tablets differs”.</p> <p>Accept an answer such as “time frame is too short since some antacids could be long-acting drugs if they contain a gelatinisation/delaying agent” but <b>not</b> just “time frame is too short since some antacids could be long-acting drugs”.</p>	2 max

Question		Answers	Notes	Total
3.	a	-21 «°C» ✓		1
3.	b	<p>28 «%» ✓</p>	<p>Accept any specific answer in the range 27 to 29 «°C».</p>	1

Question		Answers	Notes	Total
3.	c	$M_r = 94.48 \checkmark$ $\llcorner 2 \frac{(1.01 \times 2 + 16.00)}{94.48} \times 100 \Rightarrow 38.15 \llcorner \% \llcorner \checkmark$	<p><i>Award M2 only if answer is to 2 decimal places.</i></p> <p><i>Award [2] for correct final answer.</i></p> <p><i>Award [1 max] for 38.10 %.</i></p>	2
3.	d	<p>rust/corrosion «of cars and bridges»</p> <p><b>OR</b></p> <p>waste of important raw material</p> <p><b>OR</b></p> <p>soil/water salination/pollution «from run off»</p> <p><b>OR</b></p> <p>erosion of/damage to the road surface</p> <p><b>OR</b></p> <p>specific example of damage to the ecosystem</p> <p><b>OR</b></p> <p>«outdoor» temperatures may go below effective levels for NaCl «to lower freezing point» so NaCl could be wasted</p> <p><b>OR</b></p> <p>roads can refreeze causing hazards <math>\checkmark</math></p>	<p><i>Do not accept "tyre damage".</i></p> <p><i>Do not accept "economic issues" OR "environmental issues" unless specified (eg accept "increase in costs for local councils road budgets" but not "cost" alone).</i></p> <p><i>Do not accept "makes roads more slippery".</i></p>	1



**Section B**

**Option A — Materials**

Question			Answers	Notes	Total
4.	a		<p><i>Alloy:</i> mixture of <u>metal</u> with other metals/non-metals</p> <p><b>OR</b></p> <p>mixture of elements that retains the properties of a <u>metal</u> ✓</p> <p><i>Composite:</i> reinforcing phase embedded in matrix phase ✓</p>	<p><i>Award [1 max] for implying “composites only have heterogeneous/non-homogeneous compositions”.</i></p>	2
4.	b	i	<p>difference in ionic/atomic radius prevents layers sliding over each other ✓</p>	<p><i>Accept “difference in diameter/packing of cations prevents layers sliding over each other”.</i></p>	1

*(continued...)*

(Question 4b continued)

Question			Answers	Notes	Total
4.	b	ii	concern about Hg poisoning <b>OR</b> «composite» is white «so looks more like tooth» <b>OR</b> galvanic response potential exists <b>OR</b> local allergic potential <b>OR</b> less damage/destruction of healthy tooth tissue <b>OR</b> long term corrosion requires replacement <b>OR</b> gradual darkening of tooth ✓	Accept other correct responses.	1
4.	c		Any three of: sample injected into argon «plasma» ✓ atoms «of sample» are excited/ionised <b>OR</b> electrons are promoted ✓ electrons drop back/recombine with ions <b>AND</b> emit photons of characteristic energies/wavelengths/frequencies ✓ total number of photons is proportional to concentration of element ✓ actual concentration found from <u>calibration/standard</u> curve ✓	Accept "graph/plot" for "curve".	3 max

Question	Answers	Notes	Total
5.	<p><i>Any two of:</i></p> <ul style="list-style-type: none"> <li>greater selectivity ✓</li> <li>higher efficiency ✓</li> <li>longer life expectancy</li> </ul> <p><b>OR</b></p> <ul style="list-style-type: none"> <li>not easily poisoned ✓</li> <li>easier to recover ✓</li> <li>low«er» environmental impact ✓</li> <li>large range of conditions/temperatures/pressures supported ✓</li> <li>lower energy costs ✓</li> <li>increase in yield «per unit time» offsets cost of catalyst ✓</li> </ul>		<b>2 max</b>

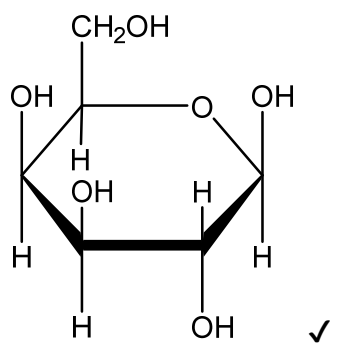
Question		Answers	Notes	Total
6.	a	$\text{Fe}(\text{CO})_5(\text{g}) \rightarrow \text{Fe}(\text{s}) + 5\text{CO}(\text{g}) \checkmark$ $2\text{CO}(\text{g}) \rightarrow \text{C}(\text{s}) + \text{CO}_2(\text{g}) \checkmark$		2
6.	b	large surface area «on which carbon nanotubes form» $\checkmark$		1
6.	c	unknown health effects <b>OR</b> unknown effect on immune systems <b>OR</b> unknown environmental effects <b>OR</b> greater inflammatory response <b>OR</b> lung damage/toxicity <b>OR</b> hazardous effect on biodiversity <b>OR</b> risk of explosion $\checkmark$	<i>Do not accept vague responses.</i>	1

Question			Answers	Notes	Total
7.	a		<p>Any two of:</p> <p>ability to form a LC phase ✓</p> <p>chemically stable ✓</p> <p>«LC phase that is» stable over suitable temperature range ✓</p> <p>polar</p> <p><b>OR</b></p> <p>being able to change orientation with applied electric field ✓</p> <p>rapid switching speed «responds to changes of voltage quickly» ✓</p>	<p>Accept “ability of molecules to transmit light under certain conditions” <b>OR</b> “rod-shaped molecules” <b>OR</b> “stable to light/not light sensitive”.</p>	2 max
7	b	i	<p>branching in LDPE prevents close packing «of chains» ✓</p> <p>LDPE is more flexible/less rigid</p> <p><b>OR</b></p> <p>LDPE has lower «tensile» strength ✓</p>	<p>Do <b>not</b> accept “difference in density”.</p> <p>Award [1 max] for stating “branching in LDPE <b>AND</b> little/no branching in HDPE”.</p>	2
7.	b	ii	<p><b>B AND</b> absence «of absorption of» C–H at 2850–3090 «cm<sup>-1</sup>»</p> <p><b>OR</b></p> <p><b>B AND</b> presence of «absorption of» C–F at 1000–1400 «cm<sup>-1</sup>» ✓</p>		1
7.	c		<p><math>(-C_2H_3Cl-)_2 (s) + 5O_2 (g) \rightarrow 4CO_2 (g) + 2H_2O (l) + 2HCl (g)</math></p> <p>correct species in reactants and products ✓</p> <p>balanced ✓</p>	<p>Accept “<math>(-C_2H_3Cl-)_2 (s) + 5.5O_2 (g) \rightarrow 4CO_2 (g) + 3H_2O (l) + Cl_2 (g)</math>”.</p> <p>Award M2 only if M1 correct.</p>	2

Option B — Biochemistry

Question			Answers	Notes	Total
8.	a	i	$C_9H_{16}O$ ✓		1
8.	a	ii	ratio of oxygen to carbon in linoleic acid lower <b>OR</b> linoleic acid less oxidized <b>OR</b> linoleic acid more reduced ✓	Accept “«average» oxidation state of carbon in linoleic acid is lower”.	1
8.	b	i	«electrophilic» addition/ $A_E$ <b>OR</b> oxidation–reduction/redox ✓		1
8.	b	ii	$\left\langle \frac{1.24 \text{ g}}{280.50 \text{ g mol}^{-1}} \Rightarrow 0.00442 \text{ «mol»} \right\rangle$ ✓ 0.00884 mol of C=C <b>OR</b> ratio of linoleic acid : iodine = 1:2 ✓ $\left\langle \text{volume of } I_2 \text{ solution} = \frac{0.00884 \text{ mol}}{0.300 \text{ mol dm}^{-3}} \Rightarrow 0.0295 \text{ «dm}^3\text{»} / 29.5 \text{ «cm}^3\text{»} \right\rangle$ ✓	Award [3] for correct final answer.	3

Question		Answers	Notes	Total
8.	c	<p>Any two of:</p> <p>increases «ratio of» HDL «to LDL» cholesterol</p> <p><b>OR</b></p> <p>decreases LDL cholesterol «level» ✓</p> <p>removes plaque from/unblocks arteries</p> <p><b>OR</b></p> <p>decreases risk of heart disease ✓</p> <p>decreases risk of stroke «in the brain» ✓</p>	<p>Accept "essential fatty acid".</p> <p>Do <b>not</b> accept "bad cholesterol" for "LDL cholesterol" <b>OR</b> "good cholesterol" for "HDL cholesterol".</p> <p>Do <b>not</b> accept general answers such as "source of energy" <b>OR</b> "forms triglycerides" <b>OR</b> "regulates permeability of cell membranes" etc.</p>	2 max

Question		Answers	Notes	Total
9.	a	<p>«reaction in which» two reactants/molecules/functional groups bond/react «to form a larger molecule/single main product» ✓</p> <p>small/tiny molecule</p> <p><b>OR</b></p> <p>H<sub>2</sub>O formed ✓</p>	<p><i>Accept formula or name of a specified small molecule other than water such as ammonia, ethanoic/acetic acid, ethanol, hydrogen sulfide etc. for M2.</i></p> <p><i>Do <b>not</b> accept just “molecule formed”.</i></p> <p><i>Award [1 max] for an example giving an equation of a condensation reaction such as the formation of a disaccharide.</i></p>	2
9.	b		<p><i>Accept “alpha” or “beta” form of galactose.</i></p>	1



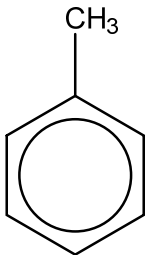
Question		Answers	Notes	Total
9.	c	<p><i>Any two of:</i></p> <p>makes the plastic more hydrophilic/water soluble ✓</p> <p>carbohydrates are broken down/hydrolysed by bacteria/microorganisms ✓</p> <p>makes plastic more accessible to bacteria as holes/channels are created</p> <p><b>OR</b></p> <p>plastic of lower density is more permeable/susceptible to water/oxygen/heat/pressure ✓</p> <p>weakens intermolecular/London/dispersion/instantaneous induced dipole-induced dipole forces «between polymer chains in the plastic» ✓</p>	<p><i>Accept “van der Waals/vdW” for “London” forces.</i></p>	<p><b>2 max</b></p>

Question		Answers	Notes	Total
10.	a	<p>«mainly» hydrocarbon/non-polar «structure» ✓</p> <p>forms London/dispersion/instantaneous induced dipole-induced dipole forces</p> <p>«with fats» ✓</p>	<p>Accept “forms van der Waals’/vdW forces”.</p> <p>Award [1 max] for “contains only one OH/hydroxyl <b>AND</b> cannot form «enough» H-bonds”.</p>	2
10.	b	<p>helps absorb calcium</p> <p><b>OR</b></p> <p>helps build bones</p> <p><b>OR</b></p> <p>helps keep bones healthy</p> <p><b>OR</b></p> <p>helps block the release of parathyroid hormone</p> <p><b>OR</b></p> <p>helps in muscle function</p> <p><b>OR</b></p> <p>helps immune system function</p> <p><b>OR</b></p> <p>cell growth</p> <p><b>OR</b></p> <p>reduction of inflammation</p> <p><b>OR</b></p> <p>protection from osteoporosis</p> <p><b>OR</b></p> <p>prevents rickets ✓</p>	<p>Accept helps prevent colon/breast/prostate cancer.</p> <p>Accept treat/prevent diabetes/heart disease/high blood pressure/multiple sclerosis.</p> <p>Accept other correct answers</p>	1

Question	Answers	Notes	Total
11.	conformation/shape altered <b>OR</b> active site altered <b>OR</b> tertiary structure altered ✓  acidic/basic/ionizable/COOH/carboxyl/NH <sub>2</sub> /amino groups in the R groups/side chains «react» ✓ exchange/lose/gain protons/H <sup>+</sup> ✓ ionic/H-bonds altered ✓	Accept “substrate doesn’t fit/fits poorly into active site” <b>OR</b> “enzyme denatures” for M1 but <b>not</b> “affects potential of enzyme to form complex with substrate”.	4

Option C — Energy

Question		Answers	Notes	Total
12.	a	$M_r(\text{C}_8\text{H}_{18}) = 114.26$ <b>AND</b> $\Delta H_c^\ominus = -5470 \text{ «kJ mol}^{-1}\text{»}$ ✓ «specific energy = $\frac{5470 \text{ kJ}}{0.11426 \text{ kg}} \Rightarrow 4.79 \times 10^4/47873/47900 \text{ «kJ kg}^{-1}\text{»}$ ✓	Award <b>[2]</b> for correct final answer.  Accept “ $48 \times 10^3 \text{ «kJ kg}^{-1}\text{»}$ ” OR “ $47.9 \times 10^3 \text{ «kJ kg}^{-1}\text{»}$ ”.	2
12.	b	wood is less useful because it requires «about three times» more mass for same energy ✓	Accept “octane is more useful because it has higher specific energy”.	1
12.	c	Any one of: wind ✓ tidal/wave ✓ hydro-electric ✓ solar ✓ thermal/geothermal ✓ plant oil ✓	Accept “biofuel/biodiesel/«bio»ethanol” but <b>not</b> just “water” or “fuel cells”.	1 max

Question		Answers	Notes	Total
13.	a	<p><math>\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3 \rightarrow \text{CH}_3\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}(\text{CH}_3)_2</math></p> <p><b>OR</b></p> <p><math>\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3 \rightarrow</math>  <math>+ 4\text{H}_2 \checkmark</math></p> <p>isomerisation/reforming/platforming/cracking <math>\checkmark</math></p> <p>Pt/Re/Rh/Pd/Ir</p> <p><b>OR</b></p> <p>catalyst <math>\checkmark</math></p>	<p><i>A structural formula is only required for the organic product, not heptane.</i></p> <p><i>Accept any correctly balanced equation showing increased branching or cyclization <b>OR</b> aromatization <b>OR</b> cracking.</i></p> <p><i>Suitable supports for catalysts may be included for M3 (eg silica, alumina, zeolite) but the symbol or name of an appropriate metal must be given (typically a noble metal). Ignore temperature and other conditions.</i></p> <p><i>Award M2 <b>AND</b> M3 for “catalytic isomerisation” <b>OR</b> “catalytic reforming” <b>OR</b> “catalytic cracking”.</i></p>	3
13.	b	<p>which specific carbon-based greenhouse gases are included</p> <p><b>OR</b></p> <p>whether non-carbon based greenhouse gases should be included</p> <p><b>OR</b></p> <p>whether CO/incomplete combustion should be included «as can be oxidized to CO<sub>2</sub>»</p> <p><b>OR</b></p> <p>how to “sum” all steps in a process creating CO<sub>2</sub></p> <p><b>OR</b></p> <p>difficult to determine both direct and indirect production of GHG/greenhouse gas emissions <math>\checkmark</math></p>	<p><i>Ignore reference to geopolitical issues (eg false recording of data by governments etc.).</i></p> <p><i>Accept “difficult to measure all sources of CO<sub>2</sub>” but <b>not</b> “difficult to measure CO<sub>2</sub> released in atmosphere”.</i></p>	1

Question		Answers	Notes	Total
13.	c	<p>Any three of:</p> <p>incoming solar radiation is short wavelength/high frequency/high energy/UV ✓</p> <p>radiated/emitted as long wavelength/low frequency/low energy/IR «radiation» ✓</p> <p>energy/IR «radiation» absorbed by «bonds in» greenhouse gases ✓</p> <p>energy radiated/emitted as IR «radiation» some of which returns back to Earth ✓</p>	<p>Do <b>not</b> accept “reflected” <b>OR</b> “bounced” <b>OR</b> “trapped”.</p>	3 max
13.	d	<p>bond length changes</p> <p><b>OR</b></p> <p>«asymmetric» stretching «of bonds»</p> <p><b>OR</b></p> <p>bond angle changes/bends</p> <p><b>OR</b></p> <p>polarity/dipole «moment» changes</p> <p><b>OR</b></p> <p>a dipole «moment» is created «when the molecule absorbs IR» ✓</p>	<p>Accept “vibration of bonds” <b>OR</b> appropriate diagram</p>	1

Question			Answers	Notes	Total
14.	a	i	<p><i>Fission: heavy nuclei <b>AND</b> Fusion: light nuclei ✓</i></p> <p><i>both increase in binding energy/energy yield «per nucleon» ✓</i></p>	<p><i>Accept “large nuclei” <b>OR</b> “greater atomic masses of nuclei” for fission <b>AND</b> “small nuclei” <b>OR</b> “smaller atomic masses of nuclei” for fusion.</i></p> <p><i>Award [1 max] for “Fission: heavy nuclei <b>AND</b> increase in binding energy «per nucleon»” <b>OR</b> “Fusion: light nuclei <b>AND</b> increase in binding energy” «per nucleon»”.</i></p>	2
14.	a	ii	<p><i>Any two of:</i></p> <p><i>no/less radioactive waste produced ✓</i></p> <p><i>abundance/low cost of fuel ✓</i></p> <p><i>larger amounts of energy released per unit mass ✓</i></p> <p><i>does not require a critical mass ✓</i></p> <p><i>can be used continuously ✓</i></p> <p><i>fusion reactor less likely to cause large-scale technological disaster ✓</i></p>	<p><i>Do <b>not</b> accept “no/less waste produced”.</i></p> <p><i>Accept “higher specific energy”.</i></p>	2 max
14.	b		<p><i>6 «hours» ✓</i></p>		1

Question		Answers	Notes	Total
15.	a	<p>«extensive» conjugation  <b>OR</b>                      alternating single and double bonds ✓</p>		1
15.	b	$  \begin{array}{c}  \text{H}_2\text{C}-\text{O}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{CH}_2)_6\text{CH}_3 \\    \\  \text{HC}-\text{O}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{CH}_2)_6\text{CH}_3 \\    \\  \text{H}_2\text{C}-\text{O}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{CH}_2)_6\text{CH}_3  \end{array}  + 3\text{CH}_3\text{OH} \longrightarrow  \begin{array}{c}  \text{H}_2\text{C}-\text{OH} \\    \\  \text{HC}-\text{OH} \\    \\  \text{H}_2\text{C}-\text{OH}  \end{array}  + 3\text{CH}_3-\text{O}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{CH}_2)_6\text{CH}_3  $ <p>ester product ✓                      glycerol <b>AND</b> correct balancing ✓</p>	<p><i>Catalyst not required for equation.</i></p> <p><i>Award M2 only if M1 is correct.</i></p>	2



Option D — Medicinal chemistry

Question		Answers	Notes	Total
16.		<p><i>Any two of:</i>  emits weak ionising radiation  <b>OR</b>  low activity/radioactivity ✓</p> <p>can be stored until material becomes inactive <b>AND</b> then disposed with normal waste ✓</p> <p>«isotopes» have short lives  <b>OR</b>  exist for a short period of time ✓</p>	<p><i>Award [1 max] for “low-level waste/LLW”.</i></p>	<p><b>2 max</b></p>

Question			Answers	Notes	Total
17.	a		prevents/interferes with the production of prostaglandins <b>OR</b> prevents/interferes with the production of substances responsible for inflammation/pain/fever ✓ at the site of injury/source of pain ✓		2
17.	b	i	react with CH <sub>3</sub> I/methyl iodide «in alkaline solution» ✓	Accept “react with CH <sub>3</sub> Cl/methyl chloride” <b>OR</b> “react with methyl halide”. Accept name or formula of a suitable specific methylating reagent (eg trimethylphenylammonium chloride etc.). Accept “hydroxy/alcohol” but <b>not</b> “hydroxide” for “hydroxyl”.	1
17.	b	ii	Any two of: interact with opioid receptors in the brain ✓  alter the structure of brain cells <b>OR</b> alter the way the brain works «so that it only works normally when the opiates are present» <b>OR</b> prevents transmission of pain impulses inside the brain ✓  release dopamine «that the person craves» <b>OR</b> give a feeling of pleasure/euphoria «that the person craves» ✓  withdrawal symptoms «prevent patient from terminating drug use» ✓	Accept specific withdrawal symptoms.	2 max

Question		Answers	Notes	Total
18.	a	in animal studies $\frac{LD50}{ED50}$ <b>AND</b> in humans $\frac{TD50}{ED50}$  <b>OR</b> in animal studies lethal dose/LD50 <b>AND</b> in humans toxic dose/TD50 ✓		1
18.	b	intravenous/IV «injection»  <b>OR</b> injection into the <u>bloodstream</u> ✓		1
19.	a	Any two of: amido ✓ ether ✓ carbonyl ✓	Accept "amide/carboxamide". Accept "alkenyl/alkene". Accept "amino/amine".	2 max
19.	b	by preventing the virus from leaving the host cell ✓ by inhibiting viral enzymes/neuraminidases «needed to release virus» ✓		2

Question		Answers	Notes	Total
20.	a	<p>blocks/binds to H2/histamine receptors «in cells of stomach lining»</p> <p><b>OR</b></p> <p>prevents histamine binding to H2/histamine receptors «and triggering acid secretion» ✓</p> <p>prevents parietal cells from releasing/producing acid ✓</p>	<p>Accept "H2-receptor antagonist/H2RA"</p> <p><b>OR</b> "blocks/inhibits action of histamine" for M1.</p>	2
20.	b	<p><b>ALTERNATIVE 1</b></p> $\text{pH} = \text{«p}K_a + \log \frac{[\text{A}^-]}{[\text{HA}]} \Rightarrow 6.35 + \log \left( \frac{0.400}{0.0200} \right) \checkmark$ <p>«pH ⇒ 7.65 ✓</p> <p><b>ALTERNATIVE 2</b></p> $K_a = 4.5 \times 10^{-7} \checkmark$ $\text{«}K_a = 0.400 \times \frac{[\text{H}^+]}{0.0200}, [\text{H}^+] \Rightarrow 2.3 \times 10^{-8} \text{ «mol dm}^{-3}\text{»}$ <p>«pH ⇒ 7.64 ✓</p>	<p>Award [2] for correct final answer.</p> <p>Do <b>not</b> accept "pH = 8".</p>	2

Question	Answers	Notes	Total
21.	<p>ring is «sterically» strained</p> <p><b>OR</b></p> <p>angles of 90° instead of 109.5/109/120° angles</p> <p><b>OR</b></p> <p>angles smaller than 109.5/109/120°/tetrahedral/trigonal planar/triangular planar angle ✓</p> <p>ring breaks up/opens/reacts «easily»</p> <p><b>OR</b></p> <p>amido/amine group «in ring» is «highly» reactive ✓</p> <p>binds to/reacts with/interferes with/inactivates <u>transpeptidase</u></p> <p><b>OR</b></p> <p>binds to/reacts with/interferes with/inactivates <u>enzyme</u> responsible for bacterial cell wall formation/cross-linking ✓</p>		3